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protons, 14 neutrons, 11 electrons (2) iii) Average mass

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number = $[85 \ 100 \times 23] + [15 \ 100 \times 25] = 23.3$ (1) iv) They both contain the same number and arrangement of electrons.
(1) v) They would have slightly different masses because $^{25}_{11}\text{Y}$ has two more neutrons than $^{23}_{11}\text{Y}$, therefore $^{25}_{11}\text{Y}$

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