

## Chapter 122 Chemical Calculations Answers

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## chapter-22-accounting-changes-and-error-analysis-multiple ...

Answers 1. a.  $2 \text{ Al} + 3 \text{ O}_2 \rightarrow 2 \text{ Al}_2\text{O}_3$  or  $2 \text{ Al} + \frac{3}{2} \text{ O}_2 \rightarrow \text{Al}_2\text{O}_3$  moles Al  $2 \text{ O}_3 = (2.3)(2)/4 = 1.2$  moles b. Using the same ratios, moles  $\text{O}_2 = (3.9)(3)/2 = 5.6$  moles 2. a. 2 moles Fe gives 3 moles  $\text{H}_2$ , moles  $\text{H}_2 = (1.7)(3)/2 = 2.6$  moles b. 3 moles  $\text{H}_2\text{SO}_4$  gives 1 mole product moles yield =  $3 \times 2.8 = 8.4$  moles 3. Mole ratios:  $2 \text{ mol Mg} / 2 \text{ mol MgO} = 1 \text{ mol Mg} : 1 \text{ mol product} : 1 \text{ mol O}$

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